

NAME:

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DATE

Atomic Dimension Worksheet AGAIN & AGAIN, & AGAIN

The Table below **contains information about several elements**. Use this table to review the concepts of atomic number, mass number, number of subatomic particles, isotopes, and charged & uncharged atoms. In each case, enough information has been provided for you to fill in all the blanks.

ELEMENT	SYMBOL	ATOMIC #	# OF p+	# OF n	# OF e ⁻	MASS #	OTHER
1. <u>Nickel</u>			<u>28</u>				
2.	<u>Al</u>				<u>10</u>		<u>Ion +3</u>
3.		<u>26</u>				<u>57</u>	
4.				<u>61</u>	<u>45</u>	<u>108</u>	
5.		<u>79</u>		<u>118</u>			<u>Neutral atom</u>
6.	<u>K</u>						<u>Ion +2</u>
7.	<u>Ti</u>				<u>24</u>		
8. <u>Calcium</u>					<u>18</u>		
9.			<u>82</u>				
10.	<u>Sn</u>					<u>120</u>	
11. <u>Platinum</u>					<u>79</u>	<u>193</u>	
12.	<u>Cu</u>	<u>30</u>				<u>65</u>	

Which of the following are CHEMICAL COMPOUNDS, MOLECULES, MIXTURES or an ATOM

13.	C	Atom	Molecule	Compound	Mixture
14.	HCl	Atom	Molecule	Compound	Mixture
15.	O ₃	Atom	Molecule	Compound	Mixture
16.	CaCO ₃ + NaCl	Atom	Molecule	Compound	Mixture
17.	Fe ₃	Atom	Molecule	Compound	Mixture
18.	HCl + H ₂ O	Atom	Molecule	Compound	Mixture
19.	Ni	Atom	Molecule	Compound	Mixture
20.	CaCO ₃	Atom	Molecule	Compound	Mixture