## **Mineral Matters Worksheet #1**

Everything in nature is composed of matter. The basic building blocks of matter are atoms. Atoms of elements combine in various ways to form compounds. Elements and compounds are the raw materials of the minerals of the earth.

Rocks are mixtures of various minerals, and minerals are made of chemical elements. Some chemical elements you may be familiar with are oxygen, silicon, calcium, sulfur, carbon, gold, and silver. There are 90 naturally occurring elements. Table 1 gives the names and symbols of some chemical elements that make up rocks and minerals.

Table 1: Common Elements and Their Symbols				
Element	Symbol	Element	Symbol	
Aluminum	Al	Magnesium	Mg	
Barium	Ba	Manganese	Mn	
Beryllium	Be	Mercury	Hg	
Boron	В	Oxygen	0	
Bromine	Br	Potassium	K	
Calcium	Ca	Silicon	Si	
Carbon	C	Silver	Ag	
Chlorine	Cl	Sodium	Na	
Copper	Cu	Sulfur	S	
Fluorine	F	Tin	Sn	
Gold	Au	Uranium	U	
Hydrogen	Н	Vanadium	V	
Iron	Fe	Zinc	Zn	
Lead	Pb	Zirconium	Zr	

Each mineral has characteristic properties, including a characteristic internal structure. Most minerals are solids at ordinary temperatures. Several examples of minerals are listed in Table 2 on the back side. Study each mineral name and its chemical formula.

In the appropriate column, write the names of the elements present in each mineral.

Table 2 Typical Minerals					
Minerals	Chemical Equation	Elements	Mineral Class		
Gold	Au				
Halite	NaCl				
Pyrite	FeS				
Cinnabar	HgS				
Marcasite	FeS <sub>2</sub>				
Quartz	SiO <sub>2</sub>				
Hematite	Fe <sub>2</sub> O <sub>3</sub>				
Fluorite	CaF <sub>2</sub>				
Calcite	CaCO <sub>3</sub>				
Feldspar	KAlSi <sub>3</sub> O <sub>8</sub>		Ÿ		
Beryl	Be <sub>3</sub> Al <sub>2</sub> (SiO <sub>6</sub> ) <sub>18</sub>				
Mica	K(Mg, Fe) <sub>3</sub> AlSi <sub>3</sub> O <sub>10</sub> (OH) <sub>2</sub>				

- 1. Which mineral is a carbonate, a compound that contains carbon and oxygen?
- 2. Which element occurs most frequently in the above Table 2.
- 3. Which two minerals contain the same elements but different numbers of atoms of each element?
- 4. Which mineral provides the liquid element mercury?
- 5. In what way are feldspar and mica chemically alike?
- 6. Can pyrite be classified as a carbonate? YES

NO

7. Explain your answer from the above question?